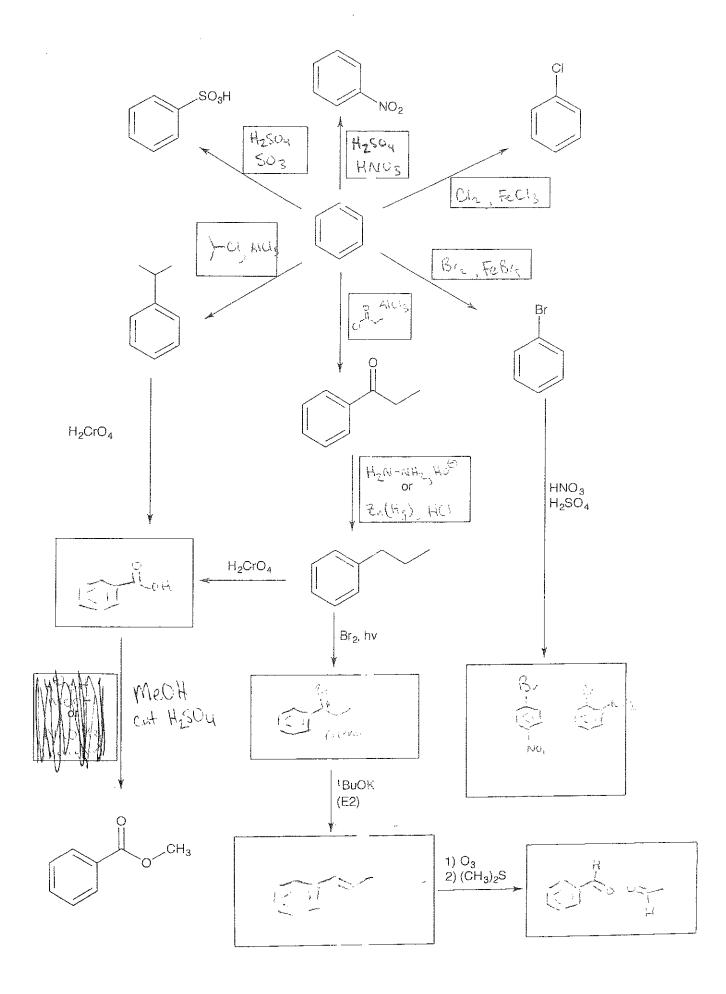
	Signature			Pg 5(24)
0	7 (4 pts each) For the	following, rank the molecu	ıles according to the dir	rections given.
	•			nder the most acidic molecule.
OH	H	OH	F ₃ C OH	H ₂ CIC OH
A	0	 }	0	U O
Q/	× 3	Žou.		<u></u>
1	under the most reactive	most reactive with nucleop molecule.		least reactive and a 4
1	CI	0	NH ₂	
77) "	Ö	0	0 0
	4	2	hware,	3
	C. Rank from least to reactive and a 4 under	most reactive with wicked still the most reactive molecule.	strong electrophiles, wit	th a I under the least
	CI	CI OCI	H_3	
		<u> </u>		3
	D. Rank from least to	nost basic, with a 1 under the	ne least basic and a 4 und	ler the most basic molecule.
	NH ₂		NH ₂	ŇH ₂
		NH ₂		
	3	A	CI	O_2N NO_2
) 1	<u></u>	<u> </u>	<u> </u>
	E. Rank from least to r	nost acidic, with a 1 under t	he least acidic and a 4 un	der the most acidic molecule.
	O II	0 0	O	O II
		0	\sim OH	н
	2	3	<u> </u>	<u> </u>
	F. Think this through!!		idic, with a 1 under the l	east acidic and a 4 under the
	most acidic molecule,	96.3	0	NO ₂ O
	NH	NH	NH	NH
				O_2N
		3	<u></u>	O
				<i>,</i>





Signature	
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Pg 10_____(34)

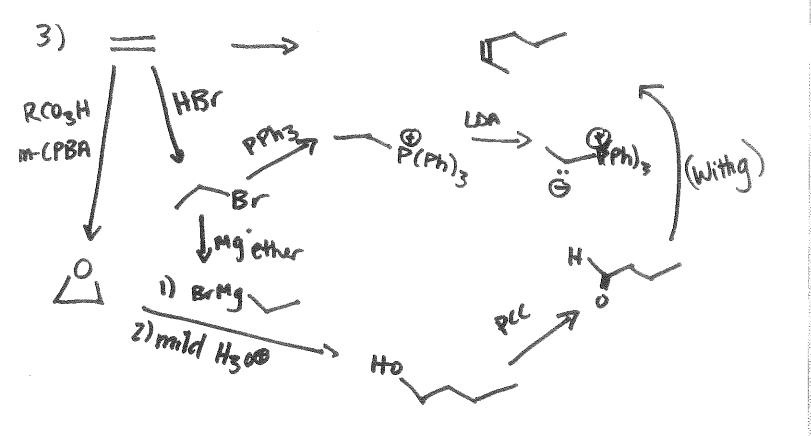
12. (34 pts.) Write the predominant product or products that will occur for each transformation. Assume each reagent only adds once to the ring. If predominantly ortho/para products are predicted, you must draw both.

) [
-Śigr	ature			

Pg	14	(14	4)
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17. Using any reagents turn the starting material into the indicated product. All the carbons in the product must come from the given starting materials. Draw all molecules synthesized along the way. When it doubt, draw the molecule!

Recognize that this one is harder than it looks because the Friedel-Crafts reaction will not work on a ring with a bad group like the nitro group on it. Therefore, the methyl group has to be made using a Wolff-Kishner or Clemmensen reduction following nitration.



- -> Zalkene: Withg!
- One combo of aldehyde + with g reagent Shown (ethyl Withg + butanal) could have made (butyl Withg + acetal).
- make ethyl with g from bromo ethane (starting material + HBr)
- -> make butanal from oxidation of 1-butanol, formed from ethyl grignard tepoxade.



